
Use of Carbon Materials as a Heating Element

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Abstract: The article presents the requirements for carbon-based materials, the graph of temperature dependence of electrical resistance, the results of tests on the study of physical and mechanical properties of heating elements made of carbon materials of different compositions.

Keywords: technology, carbon, graphite, coke, oil, coefficient of friction, cryogenic, structural, electrical, molasses, electric machines, kaolin, specific electrical resistance, density, resin, voltmeter, thermocouple, anisotropy, anisotropy

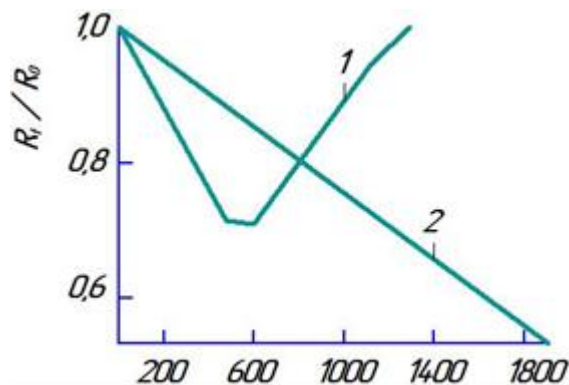
INTRODUCTION

Currently, in a number of developed countries, carbon-based materials used in various electrical equipment are produced from carbon powder materials of different origins in the technological methods of powder metallurgy [1-4]. Uglegrafite materials are used where other materials cannot be used due to their thermal conductivity, electrical conductivity, high antifriction properties and low specific electrical resistance, low coefficient of linear expansion. It depends on the physical and mechanical properties of the material used in the production of carbon-based materials from one or another type of carbon powder [5-7].

As a raw material for the production of carbon-based materials can be used any carbon-containing substances that leave a solid residue after heating. However, due to the expected physical, mechanical and operational properties of carbonaceous material, a number of requirements are set for raw materials. This is due to the fact that the chemical composition of the raw material and its physical-mechanical and technological properties have a significant impact on the formation of almost all the properties of carbonaceous materials obtained from it [8-10].

In determining the areas of application of carbon-based materials, it is necessary to determine their physical and mechanical properties. In these materials, the electrical conductivity depends on the temperature, in semiconductors the electrical conductivity decreases with increasing temperature, while in the metal it increases [11-15]. The electrical resistance of a

material depends on the temperature, for example, the electrical resistance of graphitized materials at 2000 °C decreases from 400 to 600 °C and then increases with increasing temperature [Fig. 1].



1 - graphite electrode; 2 - carbon electrode.

Figure 1. Temperature dependence of electrical resistance of carbon-based materials.

Carbon-based materials obtained on the basis of petroleum coke have an electrical resistance of 7 to 100 μO hm · m, while the resistance of graphite materials has a minimum value, and that of baked materials has a maximum value. The electrical conductivity of these materials is anisotropic. In materials obtained by pressing, this coefficient is high in the direction perpendicular to the pressure direction. The anisotropy coefficient varies from 1.2 to 2 and decreases with increasing dispersion of carbon matter particles [16 – 30].

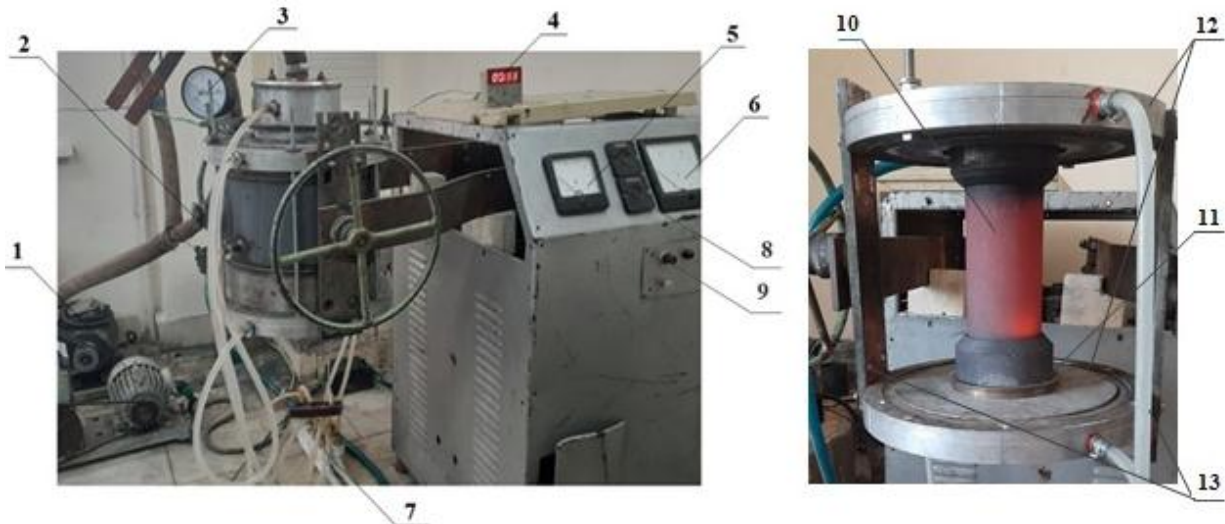
MATERIALS AND METHODS

We have prepared a carbon heating element with an outer diameter of 96 mm, an inner diameter of 76 mm and a height of 240 mm in the form of a tube based on the composition developed on the basis of local raw materials. To determine the physical and mechanical properties of the heating element, we performed tests by replacing the carbon heating element used in a special laboratory furnace. Carbon materials with different chemical compositions were prepared during the testing process. The main technical parameters of the prepared carbon materials are given in Table 1.

Table 1 Basic technical parameters of the heating element

Name of indicators	Chemical composition of the material		
	Coke + petroleum resin	Coke + molasses	Coke + molasses + kaolin
Material name			
Height, mm	240	240	240
Outer diameter, mm	96	96	96
Internal diameter, mm	76	76	76

The general view of the tested special laboratory furnace and the view of the samples placed in the furnace chamber are shown in Figure 2.



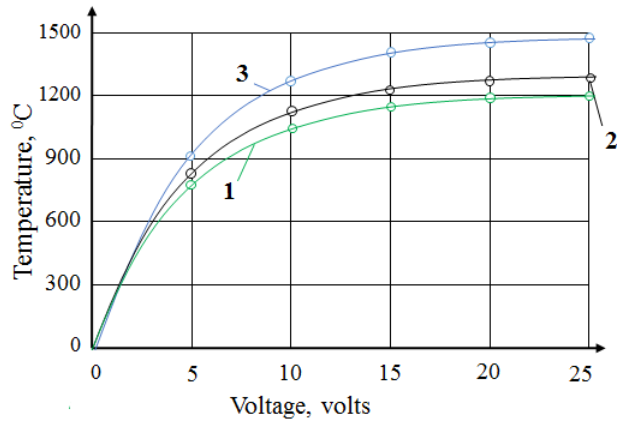
1 - vacuum pump; 2 - thermocouple; 3 - manometer; 4 - thermal sensor; 5 - ammeter; 6 - voltmeter; 7 - cooling system; 8 - automatic; 9 - thyristor block; 10 - test sample; 11 - bronze contact; 12 - radiator; 13 - current conductor.

Figure 2. Baking in a special laboratory oven and testing the physical and mechanical properties of the material

A special laboratory furnace is designed for heating samples prepared by powder metallurgy in a high vacuum or in a shielding gas environment to a temperature of 2000 °C. The outside of the oven is made of double-layered steel sheet, which is constantly cooled by passing water between the sheets to prevent the oven from overheating. To keep the heat in the furnace chamber, the outer walls of the furnace are lined with heat-resistant graphite brick. To start a special laboratory furnace: place the sample in the furnace and (1) draw the air in the furnace chamber (3) using a vacuum pump until the pressure gauge reaches a pressure of 1 MPa. An electric source is sent to the vacuum chamber through a transformer device by activating the oven start-up (8). The change of current in the special laboratory furnace (5) is adjusted by means of the thyristor block (9). The special laboratory furnace is continuously cooled with water at a rate of 0.9 m³ / h through the cooling system (7) to ensure that the contacts of the outer casing and the heating source do not overheat. The result (4), which is determined by the thermocouple of the temperature (2) in the working chamber of a special laboratory furnace, is displayed on the thermo sensor panel.

RESEARCH RESULTS

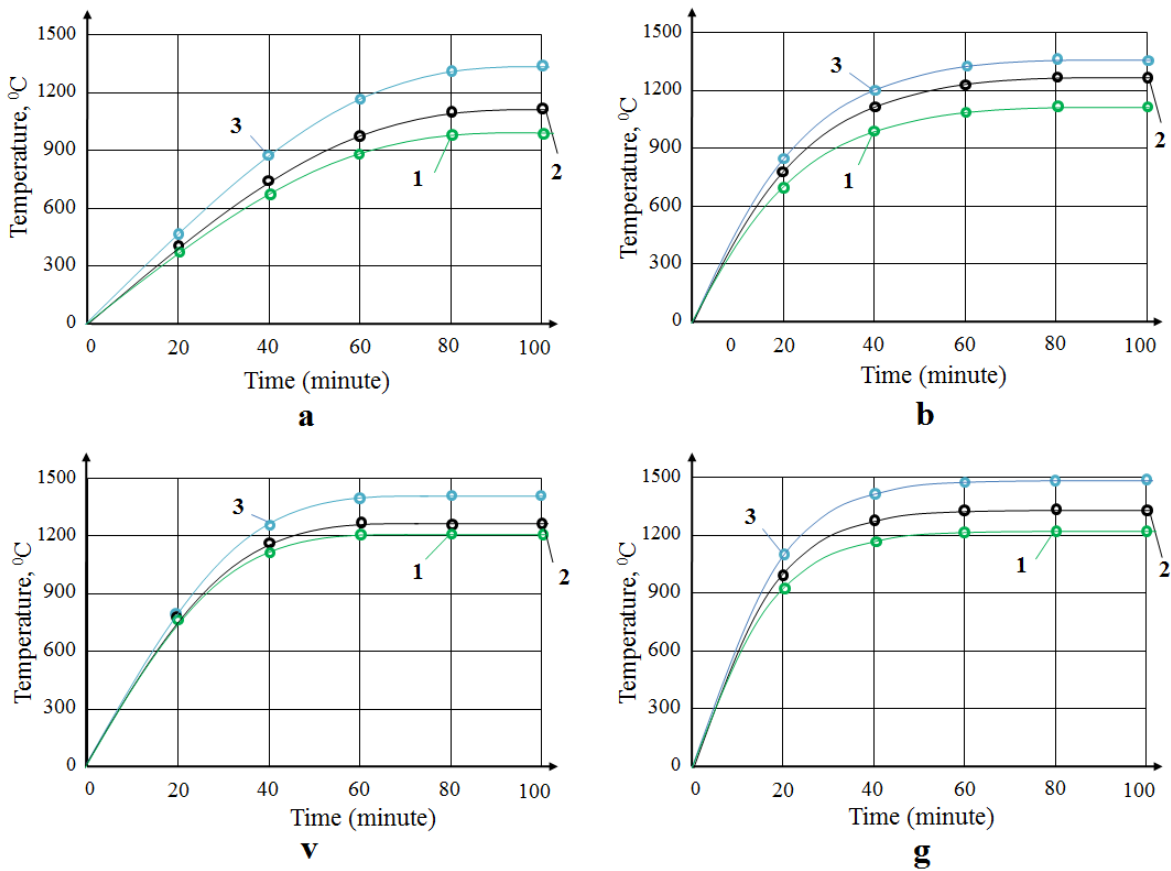
We performed the test processes by comparing their physical and mechanical properties on three samples with different compositions. Samples of different compositions were tested as a heating element. In the test, we determined the temperature of the element depending on the values of the voltage passing through the heating elements by means of the furnace voltmeter and thermocouple indicators, and the graph of their connection is shown in Figure 2



1 - Coke + oil resin; 2 - coke + molasses; 3 - coke + molasses + kaolin.

Figure 3. The change in the heating temperature of the samples depending on the voltage values passing through the heating elements

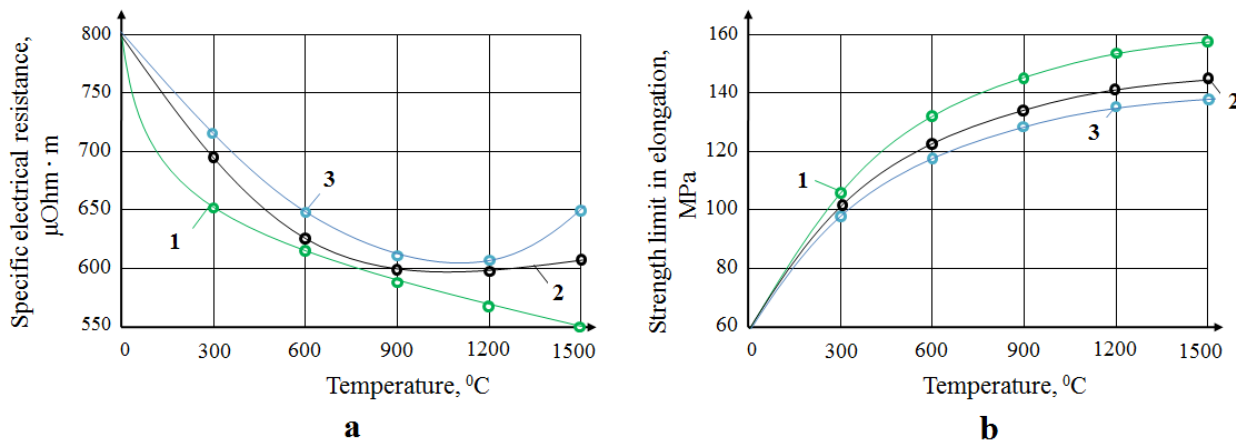
According to the test, the heating temperature of the samples also increased as the values of the electrical voltage passing through the heating elements increased. Among the test samples, a heating element containing 3 - (coke + molasses + kaolin) reached a maximum heating temperature of 1480 °C at a voltage of 25 V (Figure 3). The change in temperature over time as different voltages of voltage pass through the heating elements is shown graphically in Figure 4.



a - 10 volts; b - 15 volts; v - 20 volts; g - 25 volts.

Figure 4. Temperature change over time as different amounts of electrical voltage pass through the heating elements

According to the test results, the heating temperature of the samples also increased over time as the voltage values passing through the heating elements increased. Among the test samples, a heating element containing 3 - (coke + molasses + kaolin) was raised to a maximum temperature of 1480 °C in 60 minutes at a voltage of 25 V, and again over time the temperature value became constant (Fig. 4, g). The effect of temperature heating on the specific electrical resistance and tensile strength of the elements is shown graphically in Figure 5.



a - the specific electrical resistance; b - the tensile strength limit.

Figure 4. The effect of temperature heating on the specific electrical resistance and tensile strength of the elements

CONCLUSION

According to the test, the increase in temperature of the heating elements had a variable effect on its specific electrical resistance. In this case, the specific electrical resistance of the samples with the addition of molasses and kaolin decreased to a temperature of 1000 °C, and the electrical resistance also increased when the temperature rose from 1200 °C (Fig. 4, a).

As the temperature in the heating element increased, the tensile strength of the sample also increased. Among the test samples, a heating element containing 3 - (coke + molasses + kaolin) had a tensile strength of 156 MPa at a temperature of 1500 °C.

The results of the test analysis show that among the heating elements made of carbon materials of different composition, a sample of 3 - (coke + molasses + kaolin) has a high electrical resistance and high tensile strength under the influence of temperature and 1.2 times higher specific electrical resistance at 1500 °C. detected.

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